

5-(*p*-Tolylsulfonyl)-3-oxa-5-azatricyclo[5.2.1.0^{4,8}]-decaneCengiz Arıcı,^{a*} Dinçer Ülkü,^a
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Key indicators

Single-crystal X-ray study

T = 100 K

Mean $\sigma(\text{C}-\text{C}) = 0.004 \text{ \AA}$

R factor = 0.045

wR factor = 0.132

Data-to-parameter ratio = 15.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

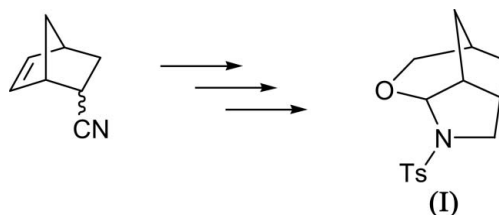
In the title compound, $\text{C}_{15}\text{H}_{19}\text{NO}_3\text{S}$, the tricyclodecane part of the molecule can be described in terms of three rings. The C_4N ring is in an envelope conformation, as is the C_5 ring. The C_5O ring adopts a boat conformation. There are no inter- or intramolecular hydrogen-bonding interactions.

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Comment

Nitrogen heterocycles, especially pyrrolidine derivatives, are widespread among both natural products and medicinally important synthetic compounds (Dogan & Garner, 2000). Bicyclo derivatives of these compounds, such as those based on the 3-azabicyclo[3.3.0]octane framework, are, however, quite rare. They are only known in a few synthetic analogues, such as prostocyclin (Malleron *et al.*, 1995) (PGI₂), antidiabetic gliclazide (Bergmeier *et al.*, 1999) and some antibacterial quinolonocarboxylic acid derivatives (Ogata *et al.*, 1991). Besides their physiological activity (Franzky *et al.*, 2000), bicyclopiperidine derivatives also serve as chiral auxiliaries (Martens & Wallbaum, 1993) in asymmetric transformations. There is also considerable interest in the development of new methods for preparing cage-like oxaheterocycles (Marchand *et al.*, 2001) and rigid amine-containing heterocycles (Becker *et al.*, 1997) (azacycles). Our studies of the synthesis of the 3-azabicyclo[3.3.0]octane framework resulted in the formation of a new tricycloaminoether, the title compound, (I). We report here the X-ray crystal structure of this interesting compound.



The tricyclodecane part of (I) can be described in terms of three rings, *A* (N/C8/C9/C13/C14), *B* (C9–C13) and *C* (O3/C11–C15). Ring *A* is in an envelope conformation, with the flap atom, C9, displaced by 0.503 (3) Å from the plane of the other four atoms. Ring *B* is also in an envelope conformation, with the flap atom, C12, lying 0.715 (3) Å from the plane of the other four atoms. Ring *C* adopts a boat conformation, in which atoms C12 and O3 are displaced from the plane through the other four atoms by 0.866 (3) and 0.515 (2) Å, respectively.

No inter- or intramolecular hydrogen-bonding interactions are present in the crystal structure of (I).

Experimental

5-(Toluene-4-sulfonyl)-3-oxa-5-azatricyclo[5.2.1.0^{4,8}]decan-2-one (100 mg, 0.326 mmol) was dissolved in tetrahydrofuran (THF; 1 ml) and added to a round-bottomed flask equipped with a magnetic stirrer, containing a 1 M solution of LiAlH₄ (10 mg, 0.261 mmol) in THF (0.2 ml) at 273 K. The reaction mixture was then stirred at room temperature for 2 h (monitored by thin-layer chromatography). A 1 N HCl solution (5 ml) and diethyl ether (5 ml) were then added and the layers were separated. The aqueous layer was extracted with diethyl ether (2 × 5 ml) and the combined organic layers were dried over Na₂SO₄ and then concentrated under reduced pressure. The crude product was purified by flash column chromatography (silica gel; hexane–EtOAc 1:5 v/v) to give 55.6 mg (58%) of the title compound, which was recrystallized from EtOH for X-ray analysis. The elemental analysis and IR and NMR spectroscopic data of (I) have already been published (Kaniskan & Dogan, 2003).

Crystal data

C ₁₅ H ₁₉ NO ₃ S	Z = 4
M _r = 293.39	D _x = 1.355 Mg m ⁻³
Monoclinic, P2 ₁ /n	Mo Kα radiation
a = 6.0285 (12) Å	μ = 0.23 mm ⁻¹
b = 16.4319 (11) Å	T = 100 (2) K
c = 14.7046 (14) Å	Block, colourless
β = 99.258 (3)°	0.25 × 0.20 × 0.15 mm
V = 1437.7 (3) Å ³	

Data collection

Enraf–Nonius CAD-4 diffractometer	2928 independent reflections
ω/2θ scans	1739 reflections with I > 2σ(I)
Absorption correction: ψ-scan (MolEN; Fair, 1990)	R _{int} = 0.022
T _{min} = 0.946, T _{max} = 0.966	θ _{max} = 26.3°
3026 measured reflections	3 standard reflections
	frequency: 120 min
	intensity decay: 0.8%

Refinement

Refinement on F ²	w = 1/[σ ² (F _o ²) + (0.0643P) ² + 0.2195P]
R[F ² > 2σ(F ²)] = 0.046	where P = (F _o ² + 2F _c ²)/3
wR(F ²) = 0.132	(Δ/σ) _{max} < 0.001
S = 1.02	Δρ _{max} = 0.22 e Å ⁻³
2776 reflections	Δρ _{min} = -0.26 e Å ⁻³
181 parameters	
H-atom parameters constrained	

H atoms were positioned geometrically and refined as riding, with C–H = 0.93–0.98 Å and U_{eq}(H) = 1.2U_{eq}(C).

Data collection: CAD-4 EXPRESS (Enraf–Nonius, 1993); cell refinement: CAD-4 EXPRESS; data reduction: CAD-4 EXPRESS; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997);

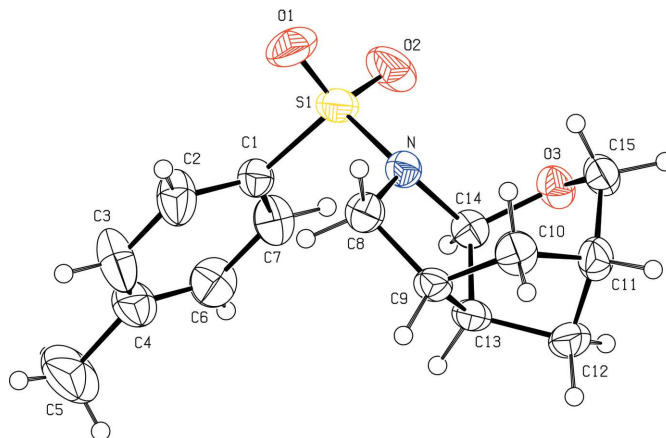


Figure 1

A plot of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as small circles of arbitrary radii.

molecular graphics: PLATON (Spek, 2003) and ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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References

- Becker, D. P., Zabrowski, D. L., Nosal, R. & Flynn, D. L. (1997). *Tetrahedron*, **53**, 1–20.
- Bergmeier, S. C., Fundy, S. L. & Punit, P. S. (1999). *Tetrahedron*, **55**, 8025–8038.
- Dogan, O. & Garner, P. (2000). *Turk. J. Chem.* **24**, 59–66.
- Enraf–Nonius (1993). *CAD-4 EXPRESS*. Version 1.1. Enraf–Nonius, Delft, The Netherlands.
- Fair, C. K. (1990). *MolEN*. Enraf–Nonius, Delft, The Netherlands.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Franzky, H., Fredricksen, S. M. & Jensen, S. R. (2000). *J. Nat. Prod.* **63**, 592–595.
- Kaniskan, H. U. & Dogan, O. (2003). *Synth. Commun.* **21**, 3833–3841.
- Malleron, J. L., Peyronel, J. F., Desmazeau, P., M'Haumadi, C. & Planiol, C. (1995). *Tetrahedron Lett.* **36**, 543–546.
- Marchand, A. P., Kumar, V. S. & Hariprakash, H. K. (2001). *J. Org. Chem.* **66**, 2072–2077.
- Martens, J. & Wallbaum, S. (1993). *Tetrahedron Asymmetry*, **4**, 637–640.
- Ogata, M., Matsumoto, H., Shimatsu, S., Kida, S., Nakai, H., Motokawa, K., Kriwa, H., Matsuura, S. & Yoshida, T. (1991). *Eur. J. Med. Chem.* **26**, 889–906.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.